

## Blaue Mahrchen Fur Alte Und Junge Kinder



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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 38 pages. Original publisher: Aberdeen Proving Ground, MD : Army Research Laboratory, 2010 OCLC Number: (OCoLC)710812456 Subject: Nitric acid. Excerpt: . . . Appendix. Mechanism Data Tables The monomethylhydrazine-red fuming nitric acid ( MMH-RFNA ) mechanism discussed in this report comprises thermochemical parameters and rate expressions for 81 species and 513 elementary reactions. Thermochemical parameters are given in table A-1, and the rate expressions are given in table A-2. The thermochemical parameters include ( 1 ) heat of formation at 298 K  $H_f(298)$ , ( 2 ) entropy at 298 K  $S(298)$ , and ( 3 ) the constant pressure heat capacity  $C_p(T)$  as a function of temperature  $T$ . For reactions whose rates (  $k$  ) are not a function of pressure, the Arrhenius expression  $k = A \exp(-E/RT)$  calculates the rate at a specified temperature from the given frequency factor (  $A$  ), temperature exponent (  $b$  ), activation energy (  $E$  ), and the ideal gas constant (  $R$  ). For reactions involving a generalized collider species (  $M$  ) ( and whose rates are therefore pressure dependent ), an effective collider concentration (  $C$  ) is calculated from the expression  $C = M p / (RT) X$ , where  $p$  is the pressure,  $X$  is the mole fraction, and  $i$  is the collider efficiency of species  $i$ . Collider efficiencies are assumed to be 1.0 unless otherwise specified. Then, one of three expressions is employed for the rate calculation. If  $T$  and  $H$  VALUE (  $S$  ) is specified, the 1 Tsang and Herron form is used and constants for that form are given,...



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